

# Life-Threatening Calcium Chloride Ingestion

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## Abstract

Calcium chloride dihydrate ( $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ ), a common component in household dehumidifiers in South Korea, poses a significant risk of toxicity upon ingestion. We present a case of life-threatening hypercalcemia following intentional calcium chloride dihydrate ingestion, with a focus on electrolyte homeostasis and physiological adaptation. An 86-year-old Korean woman presented with transient unconsciousness after ingesting dehumidifier fluid. She exhibited drowsiness and developed sinus tachycardia 10 h later. Severe hypercalcemia (19.4 mg/dL) was the main biochemical disturbance. Parathyroid hormone was initially suppressed but later rose, facilitating renal calcium excretion and phosphorus regulation. Electrolyte levels normalized by day 4. Prompt treatment, including gastric lavage and cardiac management, led to a full recovery. A rapid parathyroid hormone response played a crucial role in restoring calcium balance.

**Keywords:** Calcium chloride dihydrate; Hypercalcemia; Dehumidifiers; Deliberate ingestion; Parathyroid hormone

## Introduction

Exposure to common household or industrial chemicals, whether accidental or intentional, poses a significant risk to human health [1]. Calcium chloride ( $\text{CaCl}_2$ ) is a chemical with diverse applications, including de-icing, construction, and food production [1]. In South Korea, it is also widely used as

a dehumidifying agent in consumer products to mitigate humidity-related problems in closed environments. These dehumidifiers contain pure  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ , in concentrations ranging from 50% to 75%, and are readily available without regulatory restrictions, making them a common household item. This accessibility, however, carries a significant risk, particularly in the case of intentional ingestion, such as suicide attempts.

While isolated cases of calcium chloride poisoning have been reported, most of the existing literature focuses on localized organ damage, including gastric necrosis, metabolic acidosis, and cardiovascular complications [2-4].

This case report presents a detailed analysis of a case of severe  $\text{CaCl}_2$  intoxication, with a unique focus on the in-hospital course and its impact on systemic electrolyte homeostasis, including calcium, phosphorus, and parathyroid hormone. This case provides a rare observation of the physiological adaptation of the endocrine and renal systems to an acute, severe hypercalcemic state, offering a perspective not typically documented in previous reports.

## Case Report

### Investigations

This case report protocol was reviewed and approved by the Public Institutional Review Board of the Ministry of Health and Welfare of South Korea (approval no. P01-202501-01-027) [5].

In late October 2024, an 86-year-old Korean female (height, 158 cm; weight, 55 kg) was presented to the emergency department of our hospital with a primary complaint of altered consciousness after an intentional ingestion of a few hundred milliliters of liquid dehumidifier. The patient's family reported that the patient had ingested the liquid approximately 1 h before arrival at the hospital in a deliberate attempt to commit suicide.

Upon arrival, her vital signs were as follows: blood pressure 175/95 mm Hg, heart rate (HR) 60 bpm, and temperature 35.5 °C. She was lethargic, with no focal neurological signs, and pica behavior was not noted in her personal history.

### Diagnosis

Laboratory findings at the time of arrival (approximately 2 h post ingestion) are shown in Table 1.

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**Table 1.** Initial Laboratory Findings

| Laboratory tests                        | Results           | Reference value |
|---|-------------------|-----------------|
| CBC                                     |                   |                 |
| WBC ( $\mu\text{L}$ )                   | 18,540            | 4,000 - 10,000  |
| RBC ( $\times 10^6/\mu\text{L}$ )       | 4.59              | 4.0 - 5.5       |
| Hemoglobin (g/dL)                       | 14.3              | 12 - 16         |
| Hematocrit (%)                          | 42                | 36 - 48         |
| Platelets ( $\times 10^3/\mu\text{L}$ ) | 308               | 130 - 450       |
| Neutrophil (seg) (%)                    | 80.9              | 40 - 72         |
| Lymphocytes (%)                         | 13.9              | 17.0 - 45.0     |
| Monocytes (%)                           | 4.9               | 4 - 12          |
| Blood chemistry                         |                   |                 |
| Total protein (g/dL)                    | 7.6               | 5.7 - 8.2       |
| Albumin (g/dL)                          | 4.5               | 3.2 - 4.8       |
| Glucose (mg/dL)                         | 103               | 74 - 100        |
| BUN (mg/dL)                             | 12                | 9.0 - 26.0      |
| Creatinine (mg/dL)                      | 0.96              | 0.5 - 1.1       |
| Bilirubin, total (mg/dL)                | 1.13              | 0.3 - 1.2       |
| Aspartate transaminase (U/L)            | 30                | 8 - 34          |
| Alanine transferase (U/L)               | 19                | 10 - 45         |
| Alkaline phosphatase (U/L)              | 56                | 30 - 120        |
| Calcium, total (mg/dL)                  | 19.4 <sup>a</sup> | 8.8 - 10.6      |
| Phosphorus (mg/dL)                      | 6.6               | 2.4 - 6.1       |
| Lactate dehydrogenase (U/L)             | 252               | 140 - 271       |
| Creatine phosphokinase (U/L)            | 169               | 0 - 145         |
| C-reactive protein (mg/dL)              | 0.08              | 0.01 - 1.0      |
| Sodium (mEq/L)                          | 137               | 132 - 150       |
| Potassium (mEq/L)                       | 3.5               | 3.5 - 5.5       |
| Chloride (mmol/L)                       | 105               | 92 - 110        |

<sup>a</sup>Note the marked hypercalcemia. The blood sample was collected 2 h after  $\text{CaCl}_2$  ingestion. BUN: blood urea nitrogen; CBC: complete blood count; RBC: red cell count; WBC: white cell count.

Electrolytes were measured using an autoanalyzer (Beckman Coulter AU 689; Tokyo, Japan), and serum calcium levels were specifically assayed using the Arsenazo III method.

Chest computed tomography revealed no specific abnormalities in the lungs, pleura, or chest wall. The primary clinical manifestation upon admission was drowsiness, and the Glasgow Coma Scale score [6] decreased from 9 on hospital day 1 to 8 on hospital day 2 but subsequently improved to 13 - 14 by hospital day 4. Similarly, the Medical Research Council Muscle Strength Grading Scale score [7] declined from 5 to 3 during the first 3 days, followed by a steady recovery.

Electrocardiography revealed a normal sinus rhythm with intermittent premature ventricular beats, a QRS duration of 106 ms, and flat T waves in all leads, accompanied by a shortened QT interval ( $\text{QTc} = 395$  ms), but without J waves. Over the next 10 h, the patient's heart rate spiked, from 60 beats/min

upon admission to 130 beats/min, and a repeat electrocardiography revealed sinus tachycardia with a further shortened  $\text{QTc}$  interval (375 ms) but no changes in the QRS complex.

Sequential changes in serum calcium, phosphorus, intact parathyroid hormone, and renal excretion of calcium and phosphorus during the observation period are shown in Figure 1. The combination of severe hypercalcemia and hyperphosphatemia during the early admission period suggests a dual effect: cellular release of phosphorus due to hypercalcemia-induced cellular damage [8] and reduced renal phosphate excretion resulting from suppressed parathyroid hormone. This hypothesis is supported by elevated lactate dehydrogenase level on hospital day 2 (349 U/L; reference 140 - 271).

Ionized calcium levels were also elevated at 6.1, 6.5, and 5.3 mg/dL on hospital days 1, 4, and 6, respectively (reference range: 4.0 - 5.6 mg/dL). Magnesium levels remained within the normal range (1.80 - 2.60 mg/dL) during the observation period.

Blood gas analysis on hospital day 1 was compatible with metabolic acidosis, a finding inconsistent with calcium-alkali syndrome.

## Treatment

Given the early presentation to the emergency department, approximately 1 h after ingestion, a gastric lavage was performed using a large-bore orogastric tube (28 Fr, 9.3 mm).

The most striking electrolyte abnormality on admission was severe hypercalcemia, with a serum calcium level of 19.4 mg/dL. Extensive parenteral hydration was initiated using 2 L of normal saline. The patient's tachycardia was treated with adenosine and diltiazem, with the heart rate normalized to 60 - 70 beats/min within 48 h.

## Follow-up and outcomes

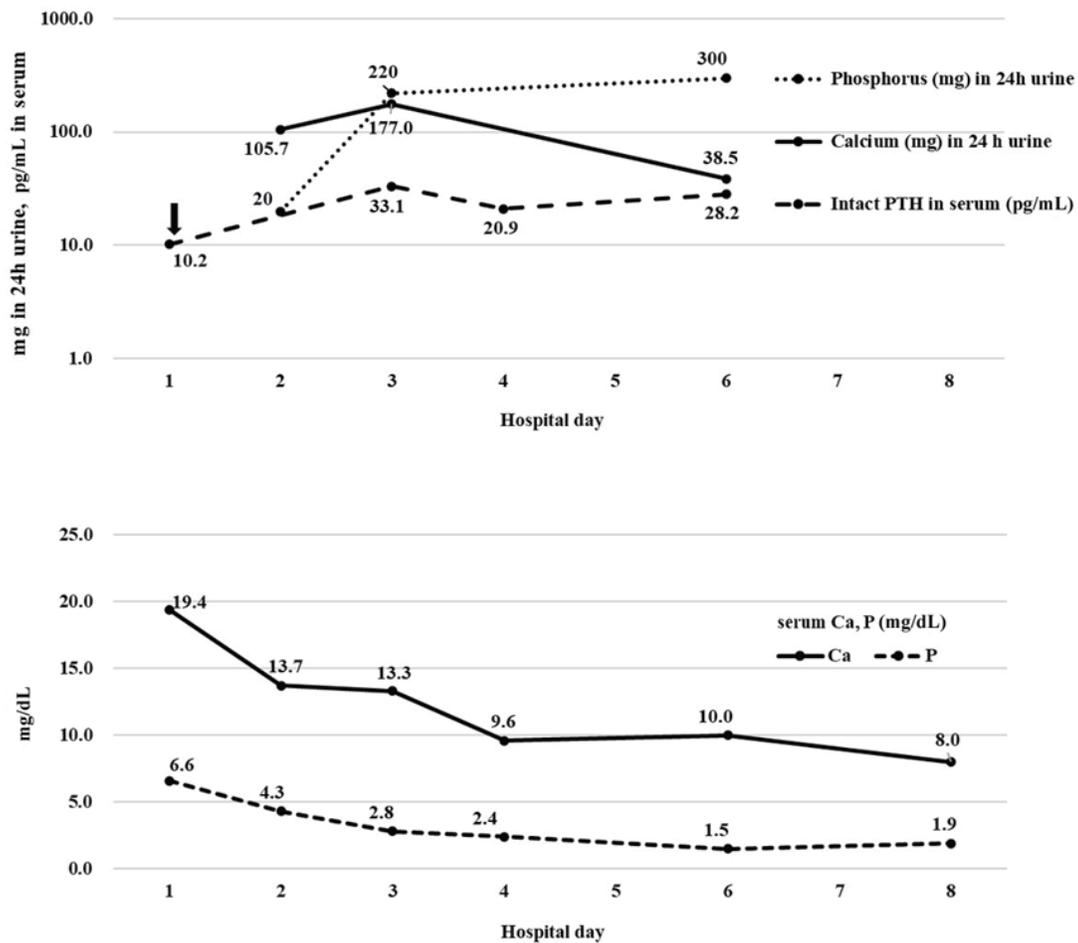
By hospital day 10, the patient reported feeling as healthy as before the ingestion of  $\text{CaCl}_2$  and was discharged. A 3-month follow-up phone call revealed that she was recovering well, with the main issue being back pain.

## Discussion

This case provides a detailed, sequential account of the physiological response to life-threatening hypercalcemia from acute  $\text{CaCl}_2$  ingestion. Unlike other case reports [2-4] that focus primarily on the corrosive and necrotic effects of this chemical, our report highlights the remarkable and rapid hormonal and renal adaptations that can facilitate homeostatic recovery.

The central observation of this case is the dynamic interplay between the parathyroid gland, serum electrolytes, and renal function, as visually demonstrated in Figure 1.

Upon admission, the patient's severely elevated serum calcium (19.4 mg/dL) led to the expected suppression of parathyroid hormone, which was initially measured at a low level



**Figure 1.** Sequential measurement of (upper; Y-axis is a logarithmic scale) serum levels of intact parathyroid hormone (pg/mL) and 24-h urine calcium and phosphorus (mg in 24-h urine), and (lower) serum calcium and phosphorus (mg/dL). Notably, the initial parathyroid hormone level, indicated by the downward arrow, was much lower than the levels measured after recovery. This result suggests that parathyroid hormone levels decreased immediately after  $\text{CaCl}_2$  ingestion, accompanied by increased renal calcium and decreased renal phosphorus excretion. In parallel, serum calcium and phosphorus levels gradually normalized by the fourth day of hospitalization. The normal reference ranges were as follows: 24-h urine phosphorus < 250 mg (females), 24-h urine calcium 400 - 1,300 mg, and serum intact parathyroid hormone 15 - 65 pg/mL. Reference values for serum calcium and phosphorus are shown in Table 1. Creatinine clearance, measured by 24 h urine on day 2, 3, and 6, was 41.7, 37.5, and 46.7 mL/min, respectively.

(10.2 pg/mL). As the systemic calcium load began to decrease following treatment, a compensatory rise in parathyroid hormone was observed, peaking at 33.1 pg/mL on hospital day 3. This hormonal response triggered enhanced renal excretion of calcium and suppressed renal excretion of phosphorus. The 24-h urine collection data clearly show this process: urinary calcium levels increased significantly from 105.7 to 177.0 mg and then decreased to 38.5 mg, while urinary phosphorus levels dropped to a nadir of 20.0 mg on hospital day 3 before rising to 300.0 mg by hospital day 6. This synchronized physiological response, even in an elderly patient with diminished physiological reserve, was highly efficient, allowing for the normalization of serum calcium and phosphorus levels by hospital day 4. This finding is significant as it demonstrates that a robust endocrine-renal axis, even in advanced age, can effectively manage an acute, severe calcium overload.

The patient's relatively benign clinical course, despite the ingestion of an estimated 300 g of  $\text{CaCl}_2$ , which far exceeds the reported lethal dose ( $\text{LD}_{50}$ ) in rats (1,000 - 1,400 mg/kg orally, European Chemicals Agency, Registration Dossier for Calcium Chloride; CAS No. 10043-52-4 [9]), is a crucial point of discussion.

Several factors contributed to this outcome. First, the patient's early presentation to the hospital, within 1 h of ingestion, allowed for prompt gastric lavage, which limited the systemic absorption of the  $\text{CaCl}_2$ . Second, the patient's surprisingly efficient hormonal and renal response was instrumental in clearing the remaining calcium from the system within a few days.

Gastric lavage is a contentious procedure in toxicology [10-12], but its role in the management of this patient warrants discussion. Gastrointestinal decontamination is a crucial intervention in South Korea, where a considerable number of

patients are affected by acute chemical intoxication from suicide attempts [13] and specific antidotes are often unavailable.

In this case, the liquid nature of the ingested solution may have made it more easily removable by the large-bore orogastric tube used. The favorable clinical outcome that followed the procedure raises the possibility that gastric lavage may be beneficial in highly selected cases with early presentation. This case illustrates the need for context-specific clinical decision-making and underscores the importance of conducting further controlled studies to clarify the role of gastric decontamination in various types of intoxication.

In addition to the endocrine and renal responses, the patient's initial laboratory results suggest a significant systemic reaction to the chemical ingestion. The complete blood count upon admission revealed an elevated white blood cell count (18,540/ $\mu$ L) and neutrophil count (80.9%) that were both well above the normal reference range (Table 1). This finding, which is often a secondary consideration in electrolyte disturbances, indicates a generalized inflammatory response, likely triggered by localized gastric mucosal injury from the concentrated  $\text{CaCl}_2$  solution. This inflammatory cascade would also contribute to the cellular damage that led to the release of cellular phosphorus, causing the initial hyperphosphatemia.

The primary limitation of this case report is that it reflects the findings of a single elderly patient, and further case studies are needed to confirm the observed physiological responses and treatment effects. Additionally, vitamin D levels were not measured during the patient's hospitalization. Despite these limitations, the detailed, real-time documentation of the patient's hormonal and renal adaptations makes this report a valuable contribution to the sparse literature on this topic.

### Learning points

The primary "take-away" lesson from this case report: Ingestion of calcium chloride-containing liquid from a dehumidifier can cause life-threatening hypercalcemia. In this case of deliberate ingestion, unconsciousness developed, and severe hypercalcemia, peaking at 19.4 mg/dL, led to suppressed parathyroid hormone level that facilitates rapid renal calcium excretion and phosphorus regulation.

Early gastric lavage and aggressive medical management contributed to the patient's survival. A robust parathyroid hormone response, even in an elderly patient, may be crucial for survival in acute  $\text{CaCl}_2$  toxicity when renal and endocrine systems are intact.

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None to declare.

### Financial Disclosure

None to declare.

### Conflict of Interest

The authors declare that they have no competing interests.

### Informed Consent

Written informed consent was obtained from the patient.

### Author Contributions

Conceptualization: SH and DK; data curation: KK and HK; formal analysis: SH; funding acquisition: JI; investigation: SH, NB, and SJ; methodology: SH; supervision: JI; validation: SH; visualization: EJ; writing - original draft: SH; writing - review and editing: SH. All authors read and approved of the final manuscript.

### Data Availability

The data supporting the findings of this study are available from the corresponding author upon reasonable request.

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